

Instructions

*Please follow the scheme of studies of relevant prospectus strictly.

*Course Code and Course Title should be carefully noted.

*Prospectus are available in soft form at university website.

*For any clash of scheme of studies in prospectus and outlines, please contact the Focal person (Department of Chemistry)

Dr. Matloob Ahmad

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M.Sc Chemistry

Semester 1

CHM-551

Physical Chemistry-I

4(3-1)

Kinetic Theory of Gases

Virial equations. Maxwells law of molecular velocities. Calculation of molecular velocities and binary collisions. Maxwell-Boltzmanns law of energy distribution.

Chemical Thermodynamics

Relation of entropy and energy with equilibrium constant and their dependence on temperature. Clausius-Clapeyron equation. Chemical potential. Partial molar quantities.

Chemical Kinetics

Integrated rate laws second and third order reactions with same and different initial concentrations of reactants. Elementary and complex reactions opposing, parallel and consecutive reactions. Steady state approximation, Lindemann theory of unimolecular reactions. Chain reactions, kinetics of interfacial reactions.

CHM-551

Practicals

- Equilibrium constant of the $KI + I_2 = KI_3$ reaction
- Kinetics of saponification of ethyl acetate
- Acid catalyzed hydrolysis of sucrose
- Study of the adsorption isotherms of acetic acid charcoal system
- Study of the charge transfer complex formation between iodine and benzene
- Determination of activation energy for the acid catalyzed hydrolysis of ethyl acetate
- Determination of partial molar volumes
- Determination of partition coefficient of a substance in two immiscible liquids.

Books Recommended:

1. R. A. Alberty, J. S. Robert, G. B. Mounqi, "Physical Chemistry". 4th Ed, John Wiley and Sons (2004).
2. D. W. Ball, "Physical Chemistry" 1st Ed, Brooks/Cole Co. Inc. (2003).
3. Engel, Thomas, P. Reid, "Thermodynamics, Statistical Thermodynamics, and Kinetics" 1st Ed, Benjamin Cummings (2006).
4. K. James, P. Wothers, "Why Chemical Reactions Happen". 5th Ed, Oxford University Press, USA (2003).
5. Smith, E. Brian, "Basic Chemical Thermodynamics" 5th Ed, Imperial College Press, (2004).
6. B. R. Stephen, S. A. Rice, J. Ross, "Physical Chemistry" 2nd Ed., Oxford University Press, USA (2000).
7. I. Chorkendorff, J. W. Niemantsverdriet, "Concepts of Modern Catalysis and Kinetics" 1st Ed, John Wiley and Sons, USA (2003).
9. J. H. Espenson, "Chemical Kinetics and Reaction Mechanism" 2nd Ed, McGraw Hill (2002).

CHM-553

Organic Chemistry-I

4(3-1)

Acids and Bases

Concepts of acids and bases; scale of acidity and basicity; pKa values; predicting acids/basis reactions from pKa values; the effect of structure on the strengths of acids and bases, field effects, resonance effects, steric effects, hydrogen bonding effects and hybridization effects, the effect of the medium on the strengths of acids and bases; the Hammett and Tafts equations, applications and limitations.

Stereochemistry

Introduction; optical isomerism; optical activity, chirality, symmetry elements and optical inactivity, relative and absolute configuration, R, S notation, methods of determining configuration. Racemic mixtures and their resolution, asymmetric synthesis, optical activity in biphenyls, alkenes and spiro compounds, stereospecific and stereoselective reactions; Geometrical isomerism. Determination of configuration of geometrical isomers, Z, E, conventions cis-and trans- isomerism in cyclic systems; Conformational isomerism conformational analysis of monosubstituted cyclohexanes, disubstituted cyclohexanes and decalin systems.

Oxidation Reduction Reactions:

a) **Oxidation:** Introduction. Oxidation of saturated, olefinic and aromatic compounds. System containing oxygen and nitrogen compounds.

b) **Reduction** Introduction. Reduction of cycloalkanes, olefins, alkynes and aromatic rings. Hydrogenolysis. Reduction of systems containing oxygen and nitrogen compounds. **CHM-553**

Practicals

Purification Techniques: Fractional distillation, fractional distillation under reduced pressure and fractional crystallization

Mixture Analysis: Analysis of two component mixture.

Books Recommended:

1. B. S. Furniss, A. J. Hannaford, P.W.G. Smith, A. R. Tatchell "Vogel's Practical Organic Chemistry", 5th Ed, Addition Wesley Longman, Harlow, England(1989).
2. J. Leonard, B. Lygo, G. Proctor, "Advanced Practical Organic Chemistry" 2nd Ed, Chapman, & Hall, London (1995).
3. H. L. Clarke, D. Hynes, "A Hand Book of Organic Analysis", Edward Arnold, London, (1995).
4. F. A Carey, R. J Sundberg, "Advanced Organic Chemistry". 3rd Ed, Part A & B, Pleman Press, New York, USA (1990).
5. K. Mislow "Stereochemistry", 2nd Ed, W. A. Benjamin Inc. New York, USA (1965).
6. E. L Eleil, S. H Wilen, L. N Mander, "Stereochemistry of Organic Compounds", 4th Ed, John Wiley & Sons, USA (1994).
7. S. H. Pine, "Organic Chemistry", 5th Ed, McGraw Hill, New York, USA (1987).
8. G. M. London, "Organic Chemistry", Addison Wesley, London, UK (1998).

CHM-555

Inorganic Chemistry-I

4(3-1)

1. BONDING MODELS FOR NON TRANSITION ELEMENTS

- (a) Covalent bond. VSEPR model followed by VBT for prediction of geometries of molecules and ions containing sigma bonds as well as pi bonds. MOT for homonuclear and heteronuclear diatomic molecules.
- (b) Metallic bond. Band theory to describe conductors, insulators and semiconductors.
- (c) 3 center 4 electrons bond, 3 center 2 electrons bond, bent bond, H bonding.

2. CHEMISTRY OF COORDINATION COMPOUNDS

Nomenclature, theories of bonding (Werners theory, Sigwick theory, Chain theory, VBT, CFT, LFT). Stereochemistry of coordination compounds, Coordination geometries (CN 2 6). Preparation of coordination compounds Stability of coordination compounds. Spectrochemical series. Application of coordination compounds in Chemistry, life and industry.

3. LANTHANIDES AND ACTINIDES

Historical survey, occurrence, separation and preparation. Oxidation states, magnetic properties of Lanthanides and Actinides. Lanthanides contraction. Applications and uses of elements and their compounds.

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Practicals

1. Separation of cations by paper chromatography: (Pb^{2+} , Cd^{2+} , Cu^{2+} , Co^{2+} , Ni^{2+} , Ag^+)
2. Preparation And Characterisation Of Complex Compounds:
(i) Sodium Cobaltinitrate (ii) Potassium trioxalato aluminate (iii) Ammonium Nickel (II) Sulphate (iv) Hexa aqua Chromium (III) chloride).
3. Complexometric Titration (Any four) Cu^{2+} / Ni^{2+} ; Ca^{2+} / Ba^{2+} ; Au^{2+} / Pb^{2+} ; Cd^{2+} / Zn^{2+} ; Ni^{2+} / Mg^{2+} ; Ca^{2+} / Zn^{2+}

Books Recommended:

1. P. Atkins, L. Jones, "Chemical Principles" 2nd Ed, Freeman and Company (2002).
2. F. Basolo, R. C. Johnson, "Coordination Chemistry: The Chemistry of Metal Complexes" W. A. Benjamin, Inc. (1964).
3. J. E. Brady, J. R. Holm, "Chemistry-The Study of Matter and Its Changes" 3rd Ed, John Wiley and Sons, Inc. (2000).
4. B. Douglas, D. McDaniel, J. Alexander, "Concepts and Models of Inorganic Chemistry" 3rd Ed John Wiley & Sons, Inc. (1994).
5. S. F. A. Kettle, "Coordination Compounds" 1st Ed, Thomas Nelson & Sons Ltd. (1969).
6. G. L. Miessler, A. T. Donald, "Inorganic Chemistry" 2nd Ed, Prentice-Hall International, Inc. Prentice-Hall, (1991).
7. D. F. Shriver, P.W. Atkins, C. H. Langford, "Inorganic Chemistry". 2nd Ed, Oxford University Press. USA (1994).

CHM-557

Environmental Chemistry

3(3-0)

Introduction, General principle and Techniques. Soil, sludge, sediment and dust analysis. Analysis of plant Material, Analysis of Atmospheric samples, Analysis of Water, determination of toxic organic Chemistry. Toxicity of heavy metals. Biological indicators, Green Chemistry, Echo toxicology.

Recommended Books:

1. B. B. Kebbekus, S. Mitra "Environmental Chemical Analysis", 1st Ed., Blackie Academic & Professional, New York, USA (1998).
2. D. Barcelo, "Environmental analysis: Techniques, Applications and Quality Assurance", Volume 13, Elsevier B.V., Netherland (1993).
3. P. Patnaik, "Handbook of Environmental Analysis", 2nd Ed, CRC Press, Taylor and Francis Group, UK (2010).

CHM-559

Analytical Chemistry-I

3(3-0)

Chemical Analysis and Data Handling

Accuracy of analytical processes such as sampling, weighing, volume measurements, precipitation, washing, filtration and ignition. Recent developments in the sampling techniques, statistical analysis; random and systematic errors, rounding off the data, arithmetic mean, median, mode, standard deviation, relative standard deviation, student t-test, F-test etc., quality control and quality assurance constructing and interpreting quality control and quality assurance constructing and interpreting quality control plots. The use of computer in data handling.

Ionic Equilibria in Solutions

Activity and activity coefficients, Hydrogen ion activity and pH for weak acids and bases, Determination of pKa and pKb value, common ion effect and its industrial applications. Buffer its composition and mechanism and buffer capacity. Stability and

formation constants of complexes, methods for their determination.

Separation Techniques

Solvent extraction Principle, factors affecting the extraction systems, Distribution la, coefficient and ratio, multiple batch extraction, practical applications in chemical analysis.

Chromatographic methods General theory of chromatography, classification of chromatographic methods, column, paper, thin-layer, and ion-exchange chromatography and their applications.

Books Recommended:

1. Blackburn, R. Thomas, "Equilibrium- A Chemistry of Solutions", 2nd Ed, Holt, Rinehart and Winston, Inc., (1969).
2. G. D. Christian, "Analytical Chemistry" 6th Ed, John Wiley & Sons, New York, USA (2003).
3. D. C. Harris, "Quantitative Chemical Analysis" 4th Ed, Freeman (1995).
4. D. A. Skoog, D. D. West, F. J. Holler, "Fundamentals of Analytical Chemistry" 6th Ed., Saunders College Publishing (1992).

CHM-561

Biochemistry-I

3(3-0)

Introduction to biochemistry, scope of biochemistry, living systems, evolution and rise of living systems, important elements of living systems including carbon, nitrogen, phosphorus, hydrogen etc. foundations of biochemistry, the physical, cellular, chemical, genetic and evolutionary foundations of life, nature of organic matter, isomerism, general reactions of different functional groups, biologically important organic compounds, carbohydrates, proteins, lipids and nucleic acids

Books Recommended:

1. D. L. Nelson, M. M. Cox. W. H. Freeman "Lehningers Principles of Biochemistry, 5th Ed. (2008).
2. D. J. Voet, G.J. Voet, C. W. Pratt "Fundamentals of Biochemistry, 3rd Ed., J. Wiley & Sons Inc. USA (2008).
3. C. K. Mathews, K. E. Van Holde, K.G. Ahern "Biochemistry, 3rd Ed., Prentice Hall (1999).

Semester 2

CHM-552

Physical Chemistry-II

4(3-1)

Electrochemistry

Theory of metallic conduction, electrode potential, electrochemical cell, electrolysis and related issues, liquid junction potential electron transfer reactions, rate of charge transfer

reaction, cell potential and thermodynamics, and Nernst equation, Voltammetry, fuel cells. Corrosion and its prevention. Ion in aqueous solution, ionic activity and Debye Huckel Theory.

Quantum Chemistry and Spectroscopy

Eigen functions and eigenvalues, Schrödinger wave equation and its applications, Hamiltonian operator, Simple harmonic oscillator. Rigid rotor, vibrator, Quantum numbers.

Symmetry Elements

Introduction, Coordinate System, Symmetry operations and symmetry Elements, The Symmetry Point Groups.

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Practicals

- Determination of molecular weight of a polymer by viscosity method
- Precipitation value of electrolytes
- Measurement of IR spectra of simple compound and their interpretation
- Measurement of cyclic voltammogram of an organic compound and its interpretation
- Determination of dipole moment of an organic liquid
- Determination of percentage composition of KMnO_4 / $\text{K}_2\text{Cr}_2\text{O}_7$ in a given solution by spectroscopy.
- Stoichiometry of a complex in solution by jobs method
- Evaluation of pKa value of indicator by spectrometric method

Books Recommended:

1. F. Cotton, Albert "Chemical Applications of Groups Theory", 1st Ed, Interscience Publishers (1963).
2. G. W. King, "Spectroscopy and Molecular Structure", 1st Ed, Rinehart and Winston (1964).
3. J. Albery, "Electrode Kinetics", 2nd Ed, Clarendon, Oxford, (1975).
4. O. M. J. Bockris, A. K. N. Reddy, "Modern Electrochemistry" 2nd Ed, Vol. I and 2, Plenum Press, New York, USA (1970).
5. D. F. Micheal, "Elements of Quantum Mechanics" 2nd Ed., Oxford University Press, USA (2005).
6. H. H. Lowell, "Group Theory and Symmetry in Chemistry" 1st Ed, McGraw Hill Book Company (1969).
7. D. H. Whiffen, "Spectroscopy" 1st Ed, Longmans Green and Co.: London, (1966).

CHM-554

Organic Chemistry-II

4(3-1)

Aliphatic nucleophilic substitution and Elimination reactions

Aliphatic nucleophilic substitution reactions Mechanisms and study of SN1, SN2, SN1,

SN₂, mechanism; neighbouring group participation intra molecular displacement by neighbouring oxygen, nitrogen, sulphur and halogen; The effects of the substrate structure, entering group, leaving group and reaction medium on the mechanisms and rates of substitution reactions.

Elimination Reactions Mechanisms study of E₁, E_{1c}B and E₂ mechanisms; attacking base, leaving group and the reaction medium on the rates and mechanisms of elimination reactions; competition between elimination and substitution reactions.

Aromatic Substitution reactions

Electrophilic substitution Aromaticity; mechanisms of substitution; orientation sulfonation, Friedel-Crafts reactions, diazo-coupling, formylation and carboxylation.

Nucleophilic substitution Mechanisms-Study of S_NAr, S_N1 and benzyne mechanisms; The effects of substrate structure, leaving group and the attacking nucleophile on the rates of substitution reactions.

Named Organic Reactions

Cannizzaro reaction, Perkin reaction, Michael reaction, Claisen-Schmidt reaction, Darzens Glycidic Ester reaction, Stobbe reaction, Mannich reaction, Wittig reaction, Ene reaction and Reformatsky reaction, Diels-Alder reaction.

CHM-554

Practicals

Organic Synthesis at least four experiments involving two step synthesis

Estimation of Amide and Carboxyl groups, Phenol and other functional groups.

Determination of Saponification value and acid value in oil.

Books Recommended:

1. F. A. Carey, R. J. Sundberg, "Advanced Organic Chemistry (Part B: Reactions and Synthesis)", 3rd Ed, Plenum Press, New York, USA (1990).
2. B. K. Carpenter, "Determination of Organic Reaction Mechanisms", "John Wiley & Sons, Inc. (1984).
3. G. R. Chatwal, "Reaction Mechanism and Reagents in Organic Chemistry", 1st Ed., Himalaya Publishing House (1987).
4. J. Fuhrhop, G. Penzlin, "Organic Synthesis Concepts, Methods, Starting Materials", 2nd Ed., Weinheim Germany (1983).
5. R. K. Mackie, D. M. Smith, "Guide book to Organic Synthesis", Longman Group Ltd. (1982).
6. J. March, "Advanced Organic Chemistry Reactions, Mechanisms and Structure", 4th Ed, John Wiley & Sons, Inc; USA (1992).
7. A. Streitwieser, C. H.H. Cock, "Introduction to Organic Chemistry", 3rd Ed, Macmillan Publishing Company (1989).
8. P. Sykes, "A Guide Book to Mechanism in Organic Chemistry", 6th Ed, Longman Group Ltd. (1986).
9. A. L. Vogel, "Elementary Practical Organic Chemistry Part III: Quantitative Organic Analysis", 1st Ed., Longman Group Ltd (1958).

1. Chemistry of Non-Aqueous Solvents;

Classification of solvents. Type of reactions in non-aqueous solvents. Physical and chemical properties of solvents. Study of reactions in liquid NH_3 , HF , SO_2 , BrF_3 , CH_3COOH and HCN . Reactions in molten salt system.

2. Pi-Acceptor Ligands:

Class of ligands. Metal carbonyls, molecular structure, localized bonding (EAN rule, 18 electron rule). Delocalized bonding (Wades rule), spectroscopic evidence of bonding situation. Chemical properties of metal-carbonyls (carbonylate anions, carbonyl hydrides and carbonyl halides). Metal nitrosyls and their derivatives. Applications of metal carbonyls and their derivatives to catalysis and organic synthesis.

3. Kinetics and mechanism of inorganic reactions:

Rate law, Stationary state approximation, Labile and inert complexes. Mechanism of substitution reactions in octahedral complexes (hydrolysis reactions, anation reactions, reactions of substituted ligand and redox reactions). Thermodynamic and kinetic stability. Half life.

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Practicals

1. Estimation Of Anions (Any four)

Chloride/Phosphate; Chloride/Nitrate; Bromide/Nitrate; Iodide/Nitrate; Borate/Acetate; Oxalate/ Chloride; Sulphate/Phosphate

2. KIO_3 Titrations (Any two)

3. Gravimetric Estimations:

Estimations of Ba^{2+} ; Oxalate ions.

Books Recommended:

1. J. E. Huheey, "Inorganic Chemistry Principles of Structure and Reactivity" 2nd Ed. Harper and Row Publishers (1978).
2. J. D. Lee, "Concise Inorganic Chemistry" 5th Ed. Chapman and Hall (1996).
3. K. M. Mackay, R. A. Mackay, W. Henderson, "Introduction to Modern Inorganic Chemistry" 5th Ed. Stanley Thorne (Publishers) Ltd. (1996).
4. G. L. Miessler, A. T. Donald, "Inorganic Chemistry". 2nd Ed, Prentice-Hall Prentice-Hall International, Inc. (1991).
5. F. A. Cotton, G. Wilkinson, "Advance Inorganic Chemistry", 5th Ed, John Wiley & Sons, Inc. (1988).
6. F. A. Cotton, G. Wilkinson, C. A. Murillo, M. Bockhmann, "Advanced Inorganic Chemistry" 6th Ed, John Wiley & Sons, Inc. USA (1999).
7. A. K. Holliday, A. G. Massey, "Inorganic Chemistry in Non-Aqueous Solvents", 6th Ed., Pergamon Press. (1985).

CHM-558

Analytical Chemistry II

3(3-0)

Properties of Light and its interaction with matter, relation between frequency , velocity and wave number, Lambert-Beer's Law and its limitations, Single and double beam

spectrophotometers, sources of light (lamp and lasers), monochromators, photomultiplier tubes, detectors, diode array and charged coupled devices, applications of UV-Vis spectrophotometer in natural product research, pharmaceutical industry, separation process, enzyme assay study, clinical studies, microbiology. Applications of IR, NMR and Mass spectrophotometer in research & development and quality control process.

Reference Books:

1. Christian, G.D. Analytical Chemistry, 6th ed., John-Wiley & Sons, New York, (2004).
2. Harris, D.C. Quantitative Chemical Analysis, 8th ed., W. H. Freeman and Company, New York, (2011).
3. Skoog, D. A., West, P.M., Holler, F.J. and Crouch, S. R., Fundamentals of Analytical Chemistry, 9th ed., Cengage Learning, (2013).
4. Braun, R.D. *Introduction to instrumental Analysis, International student Edition, (1985).*

CHM-560

Biochemistry-II

3(3-0)

Physical aspects of Biochemistry

Weak interaction in aqueous system. Ionization of water. Weak acids and weak bases. Buffers. Buffering against pH changes. Diffusion, osmosis and osmotic pressure.

Digestion Absorption and Utilization:

Carbohydrates, Lipids proteins nucleic acids, vitamins, minerals.

Enzymes

Chemical nature, nomenclature and classification. Enzyme activity. Effect of different factors on enzyme activity. Coenzymes and immobilized enzymes.

Books Recommended:

1. J. F. Robert. "Essentials of Carbohydrate Chemistry" 2nd Ed., Springer verlag, (1998).
2. D. J. Voet, G. J. Voet, C. W. Pratt "Fundamentals of Biochemistry, 3rd Ed. by. J. Wiley & Sons Inc, USA (2008).
3. S.C. Rastogi "Biochemistry, 2nd Ed, McGraw Hill, (2008).
4. S. P. Singh "Text book of Biochemistry, 4th Ed, CBS Publishers, (2008).
5. Vasudevan "Text book of Biochemistry for Medical Students, 4th Ed, J P Brothers medical (2005).

CHM-562

Introductory Spectroscopy

2(2-0)

Spectroscopy:

Introduction to principle, instrumentation and application of Electronic (UV / Visible) Atomic (Emission /Absorption), Molecular (Infrared) and Nuclear Magnetic Spectroscopy.

Recommended Books:

1. H. H. Jaffé, M. Orchin, "Theory and Applications of Ultraviolet Spectroscopy," 1st Ed, Wiley, USA (1962).
2. H. Gunther, "NMR Spectroscopy - An Introduction," 3rd Ed, John Wiley and Sons, USA (1980).
3. J. Akitt, "NMR and Chemistry; An Introduction to Nuclear Magnetic Resonance Spectroscopy," Chapman and Hall, London, (1973).
4. D. L. Pavia, G. M. Lampman, G. S. Kriz, Jr., "Introduction to Spectroscopy," 2nd Ed, W. B. Saunders, (1979).

Semester 3

Specialization in Organic Chemistry

CHM-651

Spectroscopic Organic Techniques

3(3 – 0)

(a) Introduction

Electromagnetic radiations. Wavelength, frequency, wave number and energy of electromagnetic radiations and their interconversion. Electromagnetic spectrum. Interaction transitions and spectral regions. Relaxation of the excited molecules.

(b) Ultraviolet/Visible Spectroscopy:

Introduction, Electronic transitions and absorption of electromagnetic radiations, Intensities of absorption, Beer-Lambert Law and its applications, Instrumentation and sample handling, The chromophore, Absorption by conjugated systems, Woodward fieser rules for conjugated dienes and unsaturated carbonyl systems, Absorption by aromatic compounds, Application of UV/Vis spectroscopy.

(c) Infrared Spectroscopy:

Introduction, Vibrational modes and absorption frequencies, Hooks Law, Instrumentation and sample handling, Interpretation of Infrared spectra, Characteristic absorptions frequencies of some common functional groups, Applications of Infrared spectroscopy.

(d) Nuclear Magnetic Resonance:

Introduction, Spin flipping Nuclear Precession and absorption of electromagnetic radiation, Spin relaxation, The Chemical shift and integration curve, Molecular structure and chemical shifts, Instrumentation and Sample handling, Spin splitting and coupling constants. Interpretation of NMR spectra.

(e) Mass spectrometry:

Introduction, Basic Principle, Instrumentation (theory and operation) The mass spectrum, Modes of Fragmentation of various organic molecules. Applications of mass spectrometry determination of molecular weight, molecular formula and molecular

structure. Interpretation of mass spectra.

Recommended Books:

1. H. E Duckworth, R. C Barber, V.S Barber, V.S Venkatasubramanian “Mass Spectroscopy”, 2nd Ed., Cambridge University Press, London, UK (1996).
2. E. D. Hoffmann, J. Charette, V. Stroobant, “Mass Spectrometry, Principles & Applications”, John Wiley & Sons, USA (1996).
3. A. Frigerio “Essential Aspects of Mass Spectrometry”, Spectrum Publication, Inc New York, USA (1974).
4. H. Friebolin “Basic one and two dimensional NMR Spectroscopy”, 2nd Ed, VCH (1988).
5. G. E Martin, A. S Zektzer, “Two Dimensional NMR Methods for Establishing Molecular Connectivity” VCH (1988).
6. W. Voelter “Carbon-13 NMR Spectroscopy”, 3rd Ed., VCH (1990).
7. Atta-ur-Rahman “Nuclear Magnetic Resonance Spectroscopy”, UGC, Islamabad (1989).
8. H. Gunther, “NMR Spectroscopy”, 3rd Ed., John Wiley and Sons, New York, USA (1972).
9. R. M. Silverstein, G. G. Bassler, “Spectrometric Identification of Organic Compounds” 5th Ed., John Wiley & Sons, New York, USA (1998).
10. W. Kemp, “Organic Spectroscopy”, 3rd Ed., Macmillan, London, USA (1991).

CHM-653

Rearrangements and Pericyclic Reactions

3(3 – 0)

Classification of rearrangement, Pinacol Pinacolone rearrangement, Benzil benzilic acid rearrangement, R.A involving Diazomethane, Favorski R.A, Hofman R.A. Schmidt R.A, Lossen R.A, Bayer Villiger, R.A, benzidine R.A, Fries R.A. Sigma tropic R.A. Migration of carbon, cope rearrangement, claisen rearrangement benzidine rearrangement. [1,3] H, [1,5], [1,7] H, [1,9] H migration.

Pericyclic Reactions

Conrotatory and Disrotatory motion of orbital, electrocyclic reactions, thermal cyclization, Photochemical cyclization, Hofman rule, Fukui Theory of Frontier orbitals. Introduction to cycloaddition reactions. Suprafacial and Antarafacial addition Woodward Hofman Rule. Frontier theory and mobius huckle theory for (2 + 2) and (2 + 4) thermal and photochemical cycloaddition reaction.

Recommended Books:

1. R. O .C. Norman “Principles of Organic Synthesis”, Blackie Academic & Professional, 3rd Ed. (1993).
2. F. L. Ansari, R. Qureshi and M. L. Qureshi “Electrocyclic Reactions – from Fundamentals to Research”, 1st Ed., John Wiley and Sons, (1999).
3. J. Clayden, N. Greeve, S. Warren, P. Wothers, “Organic Chemistry”, 1st Ed., Oxford University Press, USA (2001).

CHM-655

Pharmaceutical Chemistry

3(3 – 0)

Alkaloids

Introduction, occurrence, function of Alkaloids in plants, Classification, Nomenclature, Pharmaceutical Applications, Isolation, Qualitative Test and General Properties, General Method of Structure Determination. Morphines, Nicotine, Quinine.

Drugs

Introduction, Sources, Route of administration, Metabolites and mechanism of drug action. Sulfonamide, Antipyretics, Analagasic, Barbiturates, Antibiotics, their general synthesis and structure activity relationship.

Recommended Books:

1. Koji Nakanishi et “Natural Products Chemistry”, 1st Ed., Vol. I. (1974).
2. Mann, “Secondary Metabolism”, Oxford Science Publication, 2nd Ed. (1987).
3. J. D. Bu Lock “The Biosynthesis of Natural Products”, 1st Ed., McGraw-Hill, London, UK (1965).
4. S. V. Bhat, B. A. Nagasampagi, M. Sivakumar “Chemistry of Natural Product” 1st Ed., Narosa Publishing House (2005).

CHM- 657

Advanced Organic Chemistry Practicals -I

2(0-2)

Synthesis of Organic Compounds:

Students must be informed of MSDS of all compounds used in experiments.

The experiments may be arranged as per choice/requirement of instructor but should be designed from following categories;

Various experiments involving the development of amide, ester and ether linkages.

Experiments involving oxidation and reduction of organic compounds.

Synthesis of various dyes.

Recommended Books:

- 1- J. Fuhrhop, G. Penzlin, “Organic Synthesis Concepts, Methods, Starting Materials”, 2nd Ed., Weinheim Germany (1983).
- 2- A. L. Vogel, “Elementary Practical Organic Chemistry Part III: Quantitative Organic Analysis”, 1st Ed., Longman Group Ltd (1958).
- 3- F. A. Carey, R. J. Sundberg, “Advanced Organic Chemistry (Part B: Reactions and Synthesis)”, 3rd Ed, Plenum Press, New York, USA (1990).
- 4- B. S. Fumiss, A. J. Hannaford, P.W.G. Smith, A. R. Tatchell “Vogel’s Practical Organic Chemistry”, 5th Ed, Addition Wesley Longman, Harlow, England(1989).

Specialization in Analytical Chemistry

CHM-659

Electroanalytical Techniques

3(3-0)

Introduction: Electrochemistry, Electrochemical cells, Standard cell potential, Indicator electrode, Reference electrodes, Classification of electro analytical methods, Bulk method, Interfacial methods, static methods, dynamic methods

Potentiometric method of analysis: Potentiometric measurements, Indicator electrodes, Metallic indicator electrode, Membrane indicator electrode, working of potentiometer and its application including pH measurements, ion selective electrode systems, ion exchange membrane electrode, solid state membrane electrodes and bio-membrane electrodes, potentiometric titrations.

Coulometry and Electrogravimetry: Basic electrochemistry, principle, instrumentation of coulometry, principle, instrumentation of electrogravimetry, consequences of electrogravimetry, Ohmic drop, activation over potential, concentration and gas polarization, basic difference and merits/demerits of coulometry and electrogravimetry.

Reference Books:

1. Christian, G.D. Analytical Chemistry, 6th ed., John-Wiley & Sons, New York, (2004).
2. Harris, D.C. Quantitative Chemical Analysis, 8th ed., W. H. Freeman and Company, New York, (2011).
3. Skoog, D. A., West, P.M., Holler, F.J. and Crouch, S. R., Fundamentals of Analytical Chemistry, 9th ed., Cengage Learning, (2013).
4. Braun, R.D. *Introduction to instrumental Analysis, International student Edition*, (1985).

CHM-661

Advanced Separation Techniques

3(3-0)

Chromatography:

Classification of chromatographic techniques, chromatographic processes, rate theory of chromatography, Van-Deemter equation and its significance in evaluating column efficiency.

Gas Liquid Chromatography:

General principle, sample preparation/derivatization, separation process and instrumental aspects and its applications.

High Performance Liquid Chromatography:

General principle, sample preparation, separation process (normal phase and reverse phase separation), instrumentation, method development and applications.

Capillary Electrophoresis (CE):

Introduction of Electrophoresis, Theory and principle of CE, mobility, electro-osmotic flow separation by CE, instrumentation, modes of operation, applications.

Reference Books:

1. Skoog, D. A., West, P.M., Holler, F.J. and Crouch, S. R., Fundamentals of Analytical Chemistry, 9th ed., Cengage Learning, (2013).

2. Christian, G.D. Analytical Chemistry, 6th ed., John-Wiley & Sons, New York, (2004).
3. Braun, R.D. *Introduction to Chemical Analysis, International student Edition*, (1985).

CHM-663 Atomic Spectroscopy

3(3-0)

Atomic Absorption Spectrophotometry:

principle of atomic absorption spectrophotometry, concentration dependence of absorption, quantitative methodology, instrumentation for atomic absorption spectrophotometry, radiation sources, atomizers, flames, graphite furnaces and electrochemical atomizers, wavelength selectors, detectors, handling background absorption, interferences in atomic absorption spectrophotometry, sample handling in atomic absorption spectrophotometry, preparation of the sample, use of organic solvents, microwave, digestion, sample introduction methods, applications of atomic absorption spectrophotometry.

Atomic Emission Spectrophotometry:

introduction, principle of atomic emission spectrometry, atomic emission spectrometry using plasma sources, plasma and its characteristics, inductively plasma, direct current plasma, microwave induced plasma, choice of argon as plasma gas, instrumentation for ICP-MS.

Atomic Fluorescence Spectrometry:

Origin of atomic fluorescence, atomic fluorescence spectrum, types of atomic fluorescence transitions, principle of atomic fluorescence spectrometry, fluorescence intensity and analyte concentration, instrumentation for atomic fluorescence spectrometry, applications of atomic absorption spectrophotometry, interferences, merits and limitations.

Reference Books:

1. Christian, G.D. Analytical Chemistry, 6th ed., John-Wiley & Sons, New York, (2004).
2. Harris, D.C. Quantitative Chemical Analysis, 8th ed., W. H. Freeman and Company, New York, (2011).
3. Skoog, D. A., West, P.M., Holler, F.J. and Crouch, S. R., Fundamentals of Analytical Chemistry, 9th ed., Cengage Learning, (2013).
4. Braun, R.D. *Introduction to instrumental Analysis, International student Edition*, (1985).

CHM-665

Analytical Chemistry Practicals

2(0 - 2)

Practicals

The experiments may be set making use of the following instruments depending upon their. Availability, special experiments may also be designed for which a specimen list of instruments is Given below. For the innovative designing of experiments the Journal of Chemical Education may be consulted.

INSTRUMENTS

UV/Visible spectrophotometers

Flame photometers

pH-meters

Conductivity bridge

Gas chromatography

HPLC chromatography

Electro gravimetric apparatus

Atomic absorption spectrophotometer

Infrared spectrophotometers

Experiments

Determination of iron in soil by spectrophotometry.

Spectrophotometric determination of molybdate ion.

Separation of dyes using column/paper/thin layer chromatography.

Separation of sugars using paper chromatography.

Separation of amino acids using paper/thin layer chromatography.

Separation of hydrocarbons using GC/HPLC.

Determination of iron in foods products spectrophotometrically.

Determination of phosphate content in commercial fertilizers by spectrophotometry.

Determination of nickel in vegetable ghee by spectrophotometry involving solvent extraction.

Identification and spectrophotometric determination of aspirin, phenacetine and caffeine in pharmaceutical samples.

IR analysis and identification of human body stones

Mass spectrometry of mineral oil samples.

To determine pKa values for the given samples of weak acids by potentiometric method.

To determine the quality parameters i.e. pH, conductance and concentration of anions cations.

To determine Ni (II) in steel using DMG reagent by spectrophotometric method.

To determine vitamin-C concentration in the given samples.

To determine calcium and zinc in milk by atomic absorption spectrophotometer.

To determine lead in sewage sludge by atomic absorption spectrophotometer.

To determine Mn and Cr in stainless steel spectrophotometrically.

To record and characterization of IR spectra of at least 1organic compounds.

Specialization in Physical Chemistry

CHM-675 Kinetics of Complex Reactions**3(3-0)****Chemical Reactions**

Advanced theories of unimolecular reactions, Chain and non chain complex reactions, Fast reactions, Experimental techniques for measurement of fast reaction kinetics, Kinetics of catalyzed reactions

Photochemical reactions

Introduction, Photochemical reactions, photochemical reactions in gas phase and in solutions, quantum yields, flash photolysis, photochemical reaction kinetics

Interfacial Phenomena

Solid surfaces, Gas solid interfaces, thermodynamics of adsorption, adsorption at liquid surfaces, organized molecular assemblies, colloids and surfactants, liquid interfaces, surface tension and adsorption from solutions,

Recommended Books:

1. S. Asperger, "Chemical Kinetics and Inorganic Reaction Mechanisms" 2nd Ed., Springer Verlag (2003).
2. J. H. Espenson, "Chemical Kinetics and Reaction Mechanism" 2nd Ed., McGraw Hill London, UK (2002).
3. D. C. Neckers, G. von, B. Unau, W. S. Jenks, "Advances in Photochemistry", Vol. 27, John Wiley & Sons, Inc. USA (2002).
4. P. W. Atkins, "Physical Chemistry" 6th Ed, W. H. Freeman and co. New York, (1998).
5. K. J. Laidler, "The World of Physical Chemistry" 1st Ed., Oxford University Press, pp. 488 (1993).

CHM-677 Nuclear and Radiation Chemistry**3(3-0)****Nuclear Chemistry**

Introduction to Nuclear chemistry, Nuclear systematic, sources of nuclear instability, nuclear energetics , nuclear fission and fusion

Nuclear Techniques

Principles, sources of nuclear radiation. Nuclear track detectors. Etchings. Kinetics and mechanism of track etching. Nuclear materials. Nuclear techniques.

Radiation Chemistry

Tracers. Radiation Chemistry, theoretical aspects. Various models. Kinetic studies of radiolytic processes. Dosimetry (physical and chemical). Radiation chemical yields. Dose and dose rate effects on primary and secondary products. Steady state and pulse radiolysis techniques. Radiolytic studies of gaseous, water, aqueous and organic systems. Radiology.

Recommended Books:

1. G. Friedlander, J. W. Kennedy, "Nuclear and Radiochemistry", 3rd Ed. John Wiley & Sons, New York, USA (1981).
2. G. R. Choppin, J. Rayberg "Nuclear Chemistry Theory and Applications", 1st Ed., Pergaman Press, Oxford, USA (1998).
3. F. Aziz, M. A. J. Rodgers, "Radiation Chemistry Principles and Application" Ed., VCH Publishers, Inc, (1987).
4. R. Gregory, Choppin, J. Rayberg "Nuclear Chemistry Theory and Applications", 1st Ed., Pergaman Press, Oxford, USA (1998).

CHM-683

Material Chemistry

3(3-0)

Physical Chemistry of Macromolecules

Introduction, molecular forces and chemical bonding in macromolecules, configurations and conformation of polymer chains, theories of polymer solutions, spectroscopic analysis, thermal analysis, polymer rheology

Solid State

Introduction, attractive forces, properties of solids, crystal structure, crystal defects, crystallography, theories of bonding, packing of atoms in metals.

Modern Materials

Composite materials, superconductors, conducting polymers, biopolymers, Bullet proof polymers, edible plastics, smart materials, nano particles.

Recommended Books:

1. S. F. Sun, "Physical Chemistry of Macromolecules" 2nd Ed, John Wiley and Sons, INC. New York, USA (2004).
2. G. C. Bond, "Heterogeneous Catalysis" 2nd Ed., Clarendon Press. Oxford, USA (1987).
3. Anthony West "Basic Solid State Chemistry" John Wiley and sons, 1988, USA.
4. Robert J. Young, " Introduction to polymers" Capmann and Hall, 1981, USA.
5. Joel R. Fried " Polymer Science and Technology " Prentice Hall PTR. 1995.USA.
6. Fred W. Billmeyer "Text of Polymer Science" Wiley Interscience Publications, John Wiley and sons , 1984, USA.

CHM-685

Physical Chemistry Practicals

2(0 -2)

1. Sugar analysis and inversion studies by polarimetry
2. Verify Beer's Lambert's Law for the given solution.
3. Investigate the kinetics of hydrolysis of ethyl acetate in the presence of hydrochloric acid at room temperature and determination of order of reaction.
4. Interpretation of IR and NMR spectra
5. Determination of molecular weight of given sample of polymer viscometrically
6. Thermal analysis of given polymer sample with the help of available established

Recommended Books:

1. D. L. Pavia, G. M. Lampman, G. S. Kriz, Jr., "Introduction to Spectroscopy," 2nd Ed., W.B. Saunders, (1979).
2. D. W. Mathieson, "Nuclear Magnetic Resonance for organic Chemistry," Academic Press, London, UK (1967).
3. A. Douglas, F. Skoog, J. Holler, A. T. Neuman "Principles of Instrumental Analysis", 5th Ed, Saunders College Publishing, New York, USA (1997).
4. E. A. V. Ebsworth, D. W. H. Rankin, S. Craddock, "Structural Methods in Inorganic Chemistry," 2nd Ed., Blackwell, (1987).
5. E. D. Hoffmann, "Mass Spectrometry: Principles and Applications" 2nd Ed., V. Stroobant (Ed.,) John Wiley & Sons, USA (2001).
6. H. Budzikiewitz, C. Djerassi, D. H. Williams, J. R. Chapman, "Practical Organic Mass Spectrometry," John Wiley and Sons, USA (1985).

CHM-691**Organo-Transition Metal Compounds****3(3-0)**

Introduction, Cluster Compounds: Nomenclature and Structural Patterns, Metal Carbonyl Type Clusters, Anionic, Hydrido, Larger and Superlarge Carbonyl Clusters, Non-Carbonyl Clusters, Heteroatom in Clusters, Electron Counting Rules (TEC, Wades, Capping). Metal to Carbon Single, Double and Triple bonds; Acyls, Alkylidene and Alkylidyne Complexes, Bonding to Olefins, Polyolefins, Allyl, Alkyne and Arene Complexes.

Recommended Books

- F. A. Cotton, G. Wilkinson, C. A. Murillo, M. Bochmann, "Advanced Inorganic Chemistry", 6th Edition, John Wiley and Sons (2003).
- G. L. Miessler and D.A. Tarr, "Inorganic Chemistry", 3rd Edition, Pearson Education, Inc., (2004).
- W. W. Porterfield, "Inorganic Chemistry, A Unified Approach, 2nd Edition, Elsevier (1993).
- B. Douglas, D. McDaniel, J. Alexander, "Concepts and Models of Inorganic Chemistry, 3rd Edition, John Wiley and Sons (2006).

CHM-693 Inorganic Chemistry Practicals**2(0-2)****1. Conductometry**

- Titration of strong acid and weak acid with a strong base.
- Precipitation titration involving AgNO₃ and KCl.

2. Potentiometry

- Determination of K₁, K₂ and K₃ for H₃PO₄
- Determination of chloride in the presence of iodide and evaluation of AgI and AgCl

3. Spectrophotometry

- Micro determination of Cr(III) by Di-phenylcarbazide.
- Determination of Fe (II) by 1-10Phenanthroline.

- Determination of nitrites. Determination of Fe (III) by 8-hydroxyquinoline.

4. Use of some Organic Reagents for the estimation of various elements by gravimetric estimation.

- 8-Hydroxyquinoline Al (III) and Fe (III)

- Salicylaldoxime: Ni (II) in the presence of Cu (II)

- Anthranilic acid: Co (II) and Zn (II)

5. Inorganic Synthesis:

Preparation of at least six inorganic compounds/complexes in a pure state and determination of their state of purity.

Semester 4

Specialization in Analytical Chemistry

CHM-652 Thermal Methods of Analysis

3(3-0)

Thermal Analysis:

Introduction, classification and principles of thermal analysis, thermograms, instrumentations, applications and limitations of thermal analysis.

Thermogravimetric Analysis (TGA) and Derivative Thermal analysis (DTA):

Introduction and principle of thermogravimetric analysis and derivative thermal analysis, its instrumentation, applications, data interpretations, limitations.

Differential Thermal Analysis (DTA):

Introduction and principle of differential thermal analysis, its instrumentation, applications, data interpretations, limitations.

Differential Scanning Calorimetry (DSC):

Introduction and principle of differential scanning calorimetry, its instrumentation, applications, data interpretations, limitations.

Differential Photo-Calorimetry (DPC):

Introduction and principle of evolved gas analysis, its instrumentation, data interpretations, applications.

Evolved Gas Analysis (EGA):

Introduction and principle of evolved gas analysis, its instrumentation, data interpretations, applications.

Thermo-mechanical Analysis (TMA):

Introduction and principle of thermo-mechanical analysis, its instrumentation, applications, data interpretations, limitations.

Dynamic Mechanical Analysis (DMA):

Introduction and principle of dynamic mechanical analysis, its instrumentation, data interpretations, applications.

Di-electric Thermal Analysis (DETA):

Introduction and principle of di-electric thermal analysis, its instrumentation, data interpretations, applications.

Reference Books:

1. Principles of Thermal Analysis and Calorimetry, by P. J. Haines *Oakland Analytical Services, Farnhurn, Surrey, U K, Royal Chemical Society.*
2. Braun, R.D. *Introduction to Instrumental Analysis, International student Edition, (1985).*
3. Haines. P. J., Whiltby, *On Canada Mcgraw Hill Ltd., Thermal Methods of Analysis Principles, applications and problems, 1st ed. Springer, (1985).*
4. Stephen Z.D. Cheng, *Handbook of Thermal Analysis and Calorimetry, Vol. 3, Elsevier, (2002).*
5. Brown, M. E. *Introduction to Thermal Analysis: Techniques and Applications, 2nd ed., Kluwer Academic Publishers, (2001).*
6. Gabbot, P., *Principles & Applications of Thermal Analysis, Wiley-Blackwell, (2007).*

CHM-654 Nuclear Analytical Techniques

3(3-0)

Introduction to Nuclear sciences, Radioactive decay, Production of nuclear radiation, Interaction of radiation with matter, Radio-analytical techniques, Radiation detection and measurement instruments, Role of radiotracers in development of modern nuclear analytical techniques, Applications of radio-tracers in medical, environment, agriculture and industrial.

Reference Books:

1. V.S. Ramachandran, J.J. Beaudoin *Handbook of Analytical Techniques in Concrete Science and Technology, Principle, Technique and Applications. William Andrew Publishing. Norwich, New Yourk, USA, 2001.*
2. Brune, D.; Forkman, B.; Persson, B. *Nuclear analytical chemistry, Chartwell-BrattLtd., Bromley, England, United States, 1984.*
3. R Cornelis, J Caruso, H Crews, K Heumann *Handbook of elemental speciation II: species in the environment, food, medicine and occupational Health. Wiley Online Library, England, 2005*

CHM-656 Luminescence Spectrophotometry

3(3-0)

introduction, origin of fluorescence and phosphorescence spectra, Jablonski diagram, activation, deactivation, fluorescence spectrum, fluorescent and phosphorescent species,

- Educate them on prevention and control of industrial and laboratory accidents

CONTENTS

Method development and validation: Selection of analytical methods for problem solving, Optimizing the experimental procedures, Single operator characteristics, Blind analysis of standard samples, Ruggedness testing, Equivalency testing, , Sensitivity of instruments, Limits of detection and Signal-to-noise ratio.

Quality Control: Introduction and concept of quality control and quality assurance, Development of quality standards in industries, Quality control chart, Quality control in some industries, General safety practices, Good laboratory practices

Quality Assurance: Internal Methods of Quality Assessments, External Methods of Quality Assessments, Evaluation of quality assurance data, Prescriptive approach, Performance-based approach.

Automation in analytical methods; Automatic, automated and smart instruments and their applications with special emphasis on clinical, industrial and quality control aspects

Books Recommended:

1. Christian, G.D. 2003. Analytical Chemistry. Sixth edition, John Wiley and Sons, New York
2. Harvey, D. 2008. Modern Analytical Chemistry. The McGraw Hill Companies, Inc. USA.
3. Bender, G.T. 1987. "Principles of Chemical Instrumentation" W.B. Saunders Co., London
4. Hargis, L.G. 1988. "Analytical Chemistry: Printice Hall Publishers, London
5. Skoog, D.A. and J.J. Leary. 1992. "Principles of Instrumental Analysis. Saunders College Publishing Co., London
6. FAO and WHO (2000). Codex Alimentarius General Requirement Vol. 14
7. Bender, G.T. 1987. "Principles of Chemical Instrumentation" W.B. Saunders Co., London.
8. Reilley, C. 1993. Laboratory Manual of Analytical Chemistry. Allyn& Bacon, London.

CHM- 670/

Analytical Chemistry Practicals

2(0 - 2)

Practicals

The experiments may be set making use of the following instruments depending upon their. Availability, special experiments may also be designed for which a specimen list of instruments is Given below. For the innovative designing of experiments the Journal of Chemical Education may be consulted.

INSTRUMENTS

Atomic absorption spectrophotometer

Infrared spectrophotometers

GCMS

HPLC

UV/Visible spectrophotometers

Flame photometers

pH-meters

Conductivity bridge

Thin Layer Chromatography

Electro gravimetric apparatus

Experiments

Potentiometric determination of Fluoride in drinking water.
Spectrophotometric determination of Iron in soil.
Determination of pH of Hair Shampoos.
IR analysis and identification of human body stones
Ultraviolet Spectrophotometric determination of Aspirin and Caffeine in pharmaceutical samples.
Determination of iron in foods products spectrophotometrically.
Determination of Calcium by Atomic Absorption Spectrophotometry.
Determination of Mercury in Laboratory Air using Atomic Absorption Spectrophotometry.
Flame Emission Spectrometric determination of Sodium.
Qualitative and Quantitative Analysis of Fruit juices for Vitamin C using HPLC.
Enzymatic determination of Glucose in Blood.
Separation of dyes using column/paper/thin layer chromatography.
Separation of sugars using paper chromatography.
Separation of amino acids using paper/thin layer chromatography.
Identification of fingerprints by chemical test.
Analysis of Analgesics using HPLC.
Determination of phosphate content in commercial fertilizers by spectrophotometry.
Determination of nickel in vegetable ghee by spectrophotometry involving solvent extraction.
Mass spectrometry of mineral oil samples.
To determine calcium and zinc in milk by atomic absorption spectrophotometer.
Test for analysis of drugs.
To determine lead in sewage sludge by atomic absorption spectrophotometer.
Identification of fingerprints by powder test.
To record and characterization of IR spectra of at least 1 organic compounds.
Gas Chromatographic analysis of drugs and poison.
Analysis of milk, beverages and meat.

Reference Books:

1. Yolanda Picó, Chemical Analysis of Food: Techniques and Applications Academic Press, ELSEVIER, Spain, 2012.
2. Leo M. L. Nollet. Handbook of Food Analysis: Physical characterization and nutrient analysis. CRC Press, Technology & Engineering, New York USA, 2004.
3. David E. Newton, Forensic Chemistry, United States of America, (2007).

Specialization in Organic Chemistry

CHM-660

Organic Polymers

3(3 – 0)

Introduction to polymer chemistry. Step-growth polymerization, free radical addition polymerization, ionic polymerization; stereochemistry polymers; polymerization using Ziegler-Ziegler-Niegler-Natta catalyst. Stereo-regulation and conformation of polymers.

Molecular weight determination. Structure property relation. Reactions of synthetic polymers; polymers degradation and stability with special emphasis on thermal and photo-degradation.

Recommended Books:

1. W. Fred, B. Meyer “Text Book of Polymer Science”, 3rd Ed., John Wiley & Sons, (1992).
2. Joel R. Fried “Polymer Science & Technology”, Prentice Hall, Inc. (1995).
3. L.H Sperlring “Introduction to Physical Polymer Sciences”, 2nd Ed., John Wiley & Sons, USA (1990).
4. J. R. Fried “Polymer Science & Technology”, Prentice Hall, Inc. (1995).

CHM-662 Reactive Intermediates and Photochemistry 3(3-0)

Nomenclature, Preparation, Reaction of Carbene. Nitrene: Nomenclature, Preparation, Reactions, Preparation, Reaction. Arynes: Preparation, Reactions.

Photochemistry:

Introduction, Principles, Difference between thermal and photochemical reaction, laws of photochemistry, quantum yield, inter system crossing, Jablonski diagram, Photofragmentation, Norrish type I and II reaction. Photoreduction, Paterno Buchi Reaction. Reactivity of ketone, photochemistry of olefins. Pinnerization reaction.

Recommended Books:

1. N.S Isaacs “Reactive Intermediates in Organic Chemistry”, John Wiley & Sons USA (1974).
2. H. Okabe “Photochemistry of small Molecules”, John Wiley & Sons, New York, USA (1978).
3. C. W Rees, T.I. Gilchrist, “Carbenes, Nitrenes Arynes,” Nelson, London, UK (1973).

CHM – 664 Disconnection Approach 3(3-0)

The Disconnection Approach

Basic Principles: Synthesis of Aromatic Compounds, One Group: C – X Disconnections, Strategy II: Chemoselectivity, Two Group C – X Disconnections, Strategy V: Stereoselectivity A, One Group C – C Disconnections II: Carbonyl Compounds, Strategy VI: Regioselectivity, Two Group Disconnections II: 1,3-Difunctionalized Compounds and α,β -unsaturated Carbonyl Compounds, Two Group Disconnections III: 1,5-Difunctionalized Compounds, Michael Addition and Robinson Annulation, Two Group Disconnections IV: 1,2-Difunctionalized Compounds, Strategy XIII: Introduction to Ring Synthesis. Saturated Heterocycles, Three Membered Rings, Strategy XV: Use of Ketenes in Synthesis, Six-membered Rings

Recommended Books:

1. T. H. Lowry, K. S. Richardson, "Mechanism and Theory in Organic Chemistry", 3rd Ed, Harper and Row Publisher (1987).
2. G. M. Loudon "Organic Chemistry", 3rd Ed. Addison Wesley London Company (1995).
3. S. H. Pine, "Organic Chemistry", 5th Ed., McGraw Hill, New York, USA (1987).
4. G. M. Loudon, "Organic Chemistry", 2nd Ed., Addison Wesley, London (1998).
5. H.O. House "Modern Synthetic Reactions", 2nd Ed, Benjamin, California, USA (1972).

CHM – 666

Organic Catalyst and Catalysis

3(3-0)

Homogeneous and heterogeneous catalysis, Acid Catalysis, Base Catalysis, Metal ion catalysis, Hydrogenation, Asymmetric hydrogenation, Hydroboration and Hydrocyanation of olefins, Transformation of alkenes and alkynes i.e. polymerization, metathesis, dimerization and oligomerization and olefin isomerization, oxidation of olefins using catalysts, Metal complexes and Quaternary ammonium compounds in organic synthesis.

Recommended Books:

1. T. H. Lowry, K. S. Richardson "Mechanism and Theory in Organic Chemistry", 3rd Ed, Harper and Row Publisher (1987).
2. S. H. Pine, "Organic Chemistry", 5th Ed., McGraw Hill, New York, USA (1987).
3. G. M. Loudon, "Organic Chemistry", 2nd Ed., Addison Wesley, London (1998).

CHM-672

Advanced Organic Chemistry Practicals -II

2(0-2)

The experiments may be arranged as per choice/requirement of instructor but should be designed from following categories;

Synthesis of the organic compounds involving multi step synthesis using various synthetic methods. Synthesis of five or six membered heterocyclic compounds. Synthesis of targeted molecules; Anthranilic Acid. Benzilic acid, p-nitro aniline, Phenacetin and Acridon.

Recommended Books:

- 1- J. Fuhrhop, G. Penzlin, "Organic Synthesis Concepts, Methods, Starting Materials", 2nd Ed., Weinheim Germany (1983).
- 2- A. L. Vogel, "Elementary Practical Organic Chemistry Part III: Quantitative Organic Analysis", 1st Ed., Longman Group Ltd (1958).
- 3- F. A. Carey, R. J. Sundberg, "Advanced Organic Chemistry (Part B: Reactions and Synthesis)", 3rd Ed, Plenum Press, New York, USA (1990).
- 4- B. S. Furniss, A. J. Iannaford, P.W.G. Smith, A. R. Tatchell "Vogel's Practical Organic Chemistry", 5th Ed, Addition Wesley Longman, Harlow, England(1989).

Specialization in Inorganic Chemistry

CHM-676

X-ray Spectroscopy

3(3-0)

Introduction, Lattice and unit cell , geometry of crystals, crystal systems, primitive and non primitive cells, Lattice direction and planes crystal shapes Dimensional relationship, Braggs equation, reciprocal lattice, experimental methods of single & multicrystal (power) analysis, diffraction and diffractometer, identification and applications.

Recommended Books:

1. B. D. Cullity "Elements of X-ray diffraction" 2nd Ed, Addison-Wesley publishing company, California, USA (1977).
2. E. P. Bertin, "Principles and Practice of X-ray Spectrometric Analysis", Plenum Press (1975).
3. S. Prakash, G. D. Tuli, S. K. Basu, R. D. Madan, "Advanced Inorganic Chemistry" Vol.I (1997).

CHM-678

Homogenous Catalysis

3(3-0)

Reaction of CO and hydrogen, Hydroformylation, reductive carbonylation, reduction of CO by hydrogen, synthesis of water gas and shift reactions. Carbonylation reaction, Synthesis of methanol and methyl acetate, adipic ester, other carbonylation and decarbonylation reactions. Catalytic addition of molecules to C-C multiple bonds, Homogeneous hydrogenation, and hydrocylation and hydrocyanation.

Recommended Books:

1. P. Powell, "Principles of Organometallics Chemistry", 2nd Ed, London, Chapman and Hall, New York, USA (1988).
2. A. Yamamoto "Organotransition metal chemistry" John Wiley and Sons: New York, USA (1986).
3. M. Bochmann "Orgaometallics 2, complexes with transition metal carbon π -bonds" Oxford University Press, UK (1993).
4. G. L. Miessler, D. A. Tarr, "Inorganic chemistry" 2nd Ed., Prentice Hall International, USA (1998).
5. F. A. Cary, "Organic Chemistry" 7th Ed, The McGraw-Hill Company, USA (2008).

CHM-684

Radio Nuclear Chemistry

3(3-0)

Fundamentals and applied aspects of Radio activity and nuclear chemistry. Trans-Uranium elements; Natural and artificial radioactivity, methods for isotope production, nuclear reactions; mass spectrograph, Astam mass spectrograph, The structure of the nucleus; nuclear stability and radioactive decay; Types, characteristics and detection of radio active Particles; laws of radioactive decay; the interaction of radiation with matter

4. Pass. G and Sutcliffe .H., ‘Practical Inorganic Chemistry ‘. Van Nostrand Reinhold Company. 1972.

Specialization in Physical Chemistry

CHM-688 Group Theory and Solutions

3(3-0)

Advanced Group Theory

Group Algebra. Point groups. Classes Symmetry, The character table and representation, Group theory application in chemistry

Solution Chemistry

Physicochemical characteristics of solvents. Solute-solvent interaction, salvation of ions, preferential salvation. Thermodynamic methods for study of solutions

Biophysical Chemistry

Principles of biophysical chemistry; thermodynamic aspect of simple molecules, macro molecules, lipids and biological membranes; nucleic acids and proteins; enzyme kinetics and catalysis; experimental techniques

Recommended Books:

1. F. A. Cotton, “Chemical Applications of Groups Theory”, Interscience Publishers (1963).
2. A. Mohammad, “Application of Symmetry and Group Theory in Chemistry” University Grants Commission, Islamabad, (1984).
3. Alan Vincent “Molecular Symmetry and Group Theory” John Wiley & sons , 1976, USA.
4. Alberty, R. A., Robert J. S. and Mounji G. B. “Physical Chemistry”. 4th Edition , John Wiley and Sons, (2004).
5. Smith, E. Brian, “Basic Chemical Thermodynamics” 5th Edition. Imperial College Press,. (2004).
6. Stephen B. R., Rice S. A., and Ross J., “Physical Chemistry” 2nd Ed., Oxford University Press, (2000).
7. Jurg, W., “Basic Chemical Thermodynamics” W. A. Benjamin (1969).
8. Robert G. Mortimer. “Physical Chemistry” 3rd Edition, Elsevier Academic Press, UK (2008).

CHM-690 Quantum and Statistical Mechanics 3(3-0)

Statistical Mechanics

Average values, fluctuations. Partition functions of diatomic and polyatomic gases. Statistical mechanical treatment of chemical processes and equilibria. Imperfect gases, liquid state, dilute solutions and perfect crystals.

Quantum Chemistry

Operators and their properties, angular momentum. Central field problem. Variation and perturbation methods. Approximate methods in molecular quantum chemistry. Applications to quantum mechanical systems.

Theoretical and Computational Chemistry

Molecular orbital calculations. Essential concepts, semiempirical and Ab-initio methods. Reactivity. Configuration interaction method. Potential energy surfaces. Quantitative structure-activity relationship (QSAR). Computer programming and three dimensional graphics using standard packages.

Recommended Books:

1. D. J. Griffiths, "Introduction to Quantum Mechanics" 2nd Ed, Prentice Hall (2004).
2. M. G. Barrow, "Physical Chemistry" 5th Ed., McGraw Hill (1992).
3. R. Alberty, "Physical Chemistry" 17th Ed., John Wiley and Sons, USA (1987).
4. P. W. Atkins, "Physical Chemistry" 6th Ed, W. H. Freeman and co. New York, USA (1998).
5. K. J. Laidler, "The World of Physical Chemistry" 1st Ed., Oxford University Press, USA (1993).
6. K. J. Laidler, H. M. John, C. S. Bryan, "Physical Chemistry" 4th Ed., Houghton Mifflin Publishing Company Inc. (2003).
7. E. Thomas, P. Reid, "Thermodynamics, Statistical Thermodynamics", and Kinetics 1st Ed., Benjamin Cummings, (2006).

CHM-692 Advanced Spectroscopy 3(3-0)

Molecular Spectroscopy

Electromagnetic radiations, interactions of electromagnetic radiations with matter, microwave, infrared and Raman spectroscopy of polyatomic molecules, vibrational-rotational spectra,

Nuclear Magnetic Resonance

Principles of magnetic resonance. Nuclear magnetic resonance (NMR) spectroscopy. Coupling phenomenon in simple (AX_n) and complex systems. Relaxation mechanisms and their applications. Dynamic NMR. Applications in structure elucidation.

Electron Spin Resonance

Electron spin resonance spectroscopy (ESR). Principles and applications to solids and solutions.

Recommended Books:

1. J. D. Graybal, "Molecular Spectroscopy," McGraw-Hill, New York, USA (1988).
2. G. M. Barrow, "Introduction to Molecular Spectroscopy," 2nd Ed, McGraw-Hill,

7. Determination of critical micelle concentration of selected surfactants in aqueous and non-aqueous media by surface tension and conductivity.
8. Determination of free energy of micellization of selected surfactants in aqueous and non-aqueous media by surface tension and conductivity.
9. Determination of equivalence point of acid-base titration by electrical conductivity.
10. Determination of degree of dissociation of weak electrolytes.
11. Determination of pK_a values of acids.
12. Determination of pK_a values of indicators.
13. Preparation of buffers of required pH values.

Books recommended:

1. C.W. Garland, J.W. Nibler and DP Shoemaker, Experiments in Physical Chemistry, Mc Grawhill, 7th edition (1996).
2. James, A. M., Prichard, F. E., *Practical Physical Chemistry*, 3rd ed., Longman Group Limited, New York, (1974).
3. A. Findly's Practical Physical Chemistry, Longmann, London(1972).
4. LP Gold, L. Gold, Physical Chemistry Laboratory, Primis Publishers (1997) ISBN: 0072902698.